An Unusual Ring-enlargement in the Reaction of Acetylenic Esters and Ketones with N,N-Dimethylaminoallene

Willem Klop and Lambert Brandsma

Organisch Chemisch Laboratorium, Rijksuniversiteit Utrecht, Croesestraat 79, 3522 AD Utrecht, The Netherlands

The four-membered ring compounds formed from N,N-dimethylaminoallene and acetylenic esters or ketones are converted into derivatives of 2,3-dihydroazepine upon heating.

In the course of our study of the reactivity of allenic amines $H_2C=C=CH-NR_2^1$ we investigated the reaction between N,N-dimethylaminoallene (1) and acetylenic esters (2). Slow

addition at 50 °C of (1) to a twofold excess of methyl but-2-ynoate afforded, after removal of the excess of acetylenic ester by distillation, the expected four-membered ring compound

Scheme 1

R = H, X = COMe or $COPr^n$

(3) in good yield. When the same procedure was carried out at -40 °C for methyl propiolate, heating during distillation of (3) gave rise to a different compound. The structure of

this compound was shown by ¹H and ¹³C n.m.r., i.r., and mass spectroscopy to be (6).† Heating a solution of (3; R=Me) in tetrachloroethene under reflux (121 °C) for 3 h, led to complete conversion into (6). On cooling the solution obtained after conversion (at 105 °C) of 25% of the starting compound (3), the allenic derivative (4) was detected in the mixture; a maximal accumulation of 10% could be attained. The presence of the allene was detected in the i.r. (C=C=C at 1957 cm⁻¹) and ¹H n.m.r. spectra. Only the *E*-isomer was present. Similar results were obtained for acetylenic *ketones*. The ring opening took place readily at room temperature and upon heating (66 °C) the allenic compound (4) closed smoothly to give the seven-membered ring (6). On the basis of these observations we propose the reaction course shown in Scheme

The relatively easy opening of the four-membered ring (3), may be explained by the simultaneous presence of the pushing NMe₂ and the pulling substituent X ('push-pull system'). Closely analogous systems, lacking either the NMe₂ (3; $X = CO_2Me_3$, H in place of NMe₂) or the appropriate X group (3; $X = CH_2OH$), did not open on heating for 2—3 h at 120 °C.

The allenic compound (4), formed by ring opening of (3) can be written as the mesomeric zwitterion structure (4a), in which one of the methyl protons is transferred to the negatively charged carbon atom of the original allenic system. The ylide structure (5) finally undergoes a cyclization with charge neutralization and formation of (6).

To our knowledge this unusual ring transformation has no related precedents.

Received, 7th June 1983; Com. 740

Reference

 H. D. Verkruijsse, H. J. T. Bos, L. J. de Noten, and L. Brandsma, Recl. Trav. Chim. Pays-Bas, 1981, 100, 244.

[†] The 1 H n.m.r. spectra of (6) showed signals, *inter alia*, at δ 7.3 (7-H), 6.4—6.5 (5-H), and 5.3—5.5 (4-H).